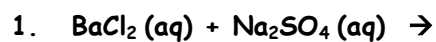


Name _____ Date _____ Period _____

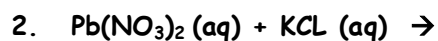


Chemical Reactions Ws #4: Double Replacement Reactions

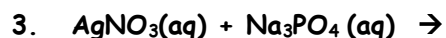
For each reaction write the (a) Molecular, (b) Complete Ionic, and (c) Net Ionic Equation.



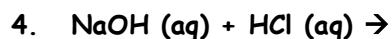
a.	Molecular Equation	
b.	Complete Ionic Equation	
c.	Net Ionic Equation	



a.	Molecular Equation	
b.	Complete Ionic Equation	
c.	Net Ionic Equation	



a.	Molecular Equation	
b.	Complete Ionic Equation	
c.	Net Ionic Equation	



a.	Molecular Equation	
b.	Complete Ionic Equation	
c.	Net Ionic Equation	



Write the balanced equation and the net ionic equation for each of the following solution mixtures. If neither water or a precipitate forms, write "No Rxn"

- Ammonium Sulfate and Barium Nitrate
- Nitric Acid and Potassium Hydroxide
- Lead(II) Nitrate and sodium chloride
- Sodium Phosphate and Potassium nitrate
- Hydrochloric Acid and Barium Hydroxide
- Copper(II) chloride and Sodium Hydroxide